

AVIATION AND AERONAUTICAL ENGINEERING



Italian Naval Air Station
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Number 2

SPECIAL FEATURES

EFFECT OF INITIAL STRESS OF REDUNDANT TRUSS MEMBERS
SOME OUTSTANDING PROBLEMS IN AERONAUTICS
THE 180 H.P. MERCEDES AERO-ENGINE
GERMAN CONCEPTIONS IN AIRPLANE CONSTRUCTION
PERMEABILITY OF BALLOON FABRICS

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these things
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wings and wings
Dewey wings that close
and stir
And yet more wings and
more and more

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AUGUST 15, 1918

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INDEX TO CONTENTS

	PAGE		PAGE
Effect of Initial Stress of Redundant Truss Members	101	Permeability of Balloon Fabrics	105
Some Outstanding Problems of Aerodynamics	104	Where Liberty Aircoaches Are Made	107
The 180 H. P. Mercedes Aero-Engine	106	Digest of Foreign Aeronautical Press	108
German Conceptions in Airplane Construction	110	News of the Tomorrow	110

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No. 2

Effect of Initial Stress on Redundant Truss Members

By H. T. Booth, B.S., M.S.

It often occurs, in the stress analysis of a wire truss, that two or more members are supplied to absorb the load at a given point. The solution for the stress is statically indeterminate.

If there is initial stress in any of the members, the problem becomes more complicated, but is comparatively readily solved by the same theory. The methods, as applied to airplane structures, have not, heretofore, been discussed in current technical papers, and it is the writer's belief that a brief analysis of the truss stress involved in such a structure will be of interest, and possibly of some assistance to airplane designers.

Suppose we have to analyze a portal structure such as shown in Fig. 1. The top, bottom and side members are, be assumed to develop little or no strain under the imposed load, since they are, in a rule, either composed of heavy heavy member material, or partially wood and the remainder steel. Also the left-hand side is fixed.

Then under any vertical load P imposed upon the frame at point A , a deformation will occur which produces a change in the lengths of members I and II , and consequently a resultant stress. The problem is to determine the relation between the stresses in I and II and the load P , expressed in such a form that the stresses may be computed.

In the following analysis, let

(1) P = vertical stress and stress

(2) P = vertical stress in member I

(3) P = vertical stress in member II

(4) P = length of I corresponding to S_1

(5) P = length of II corresponding to S_2

(6) P = modulus of elasticity of the material

(7) P = area of member in square inches

(8) P = stress in I or II due to load P

(9) P = total stress in member I or II

(10) P = length of member under stress S

The length l of either of the two diagonals (which we shall assume to be under negligible of supporting a resultant compressive stress) for any stress S , may be expressed by the two equations:

$$l = \frac{S}{E} + l_0 \quad (1)$$

$$l = \frac{S}{E} + l_0 \quad (2)$$

The deformation of stress of the member is

$$S_1 = (l_1 - l_0) \frac{E}{L} \quad (3)$$

$$S_2 = (l_2 - l_0) \frac{E}{L} \quad (4)$$

By our assumption that the stress in the longitudinal and vertical members, produced by the load P , may be neglected in comparison with the stress in the diagonals by the same load P , we obtain another equation of condition, namely,

$$S_1 = S_2 \quad (5)$$

where l is the vertical deflection of the free ends of the diagonals.

Introducing these vertical deflections along their respective diagonals,

$$S_1 = S_2 \sin \alpha_1 \quad (6)$$

$$S_1 = S_2 \sin \alpha_2 \quad (7)$$

By equation (5)

$$S_1 \sin \alpha_1 = S_2 \sin \alpha_2$$

From equations (3) and (4), if the two diagonals are similar, and the frame is a parallelogram,

$$\frac{dS_1}{dS_2} = \frac{dS_1}{dS_2} \quad (8)$$

or the rate change of the stress in member I is numerically equal to, but opposite in sign to the rate change of the stress in member II , both being due to a change in the load P .

The above equation affords a physical idea of the stress action involved in a change of the load P , but as yet we have developed no explicit relation between P and the stress in either diagonal. The next step will be to determine expressions by which the exact stress in either member may be computed.

We know that the sum of the vertical components of the induced stress in the diagonals must equal the vertical load P , so

$$P = S_1 \sin \alpha_1 + S_2 \sin \alpha_2 \quad (9)$$

$$P = S_1 \sin \alpha_1 + S_2 \sin \alpha_2 \quad (10)$$

$$S_1 = S_2 \frac{\sin \alpha_2}{\sin \alpha_1}$$

$$S_2 = S_1 \frac{\sin \alpha_1}{\sin \alpha_2}$$

Since

$$S_1 = S_2 \frac{\sin \alpha_2}{\sin \alpha_1} \left(\frac{S_1}{S_2} + 1 \right) \quad (11)$$

$$S_2 = S_1 \frac{\sin \alpha_1}{\sin \alpha_2} \left(\frac{S_2}{S_1} + 1 \right) \quad (12)$$

Integrating and solving for S_1

$$S_1 = \frac{P}{\left(\frac{\sin \alpha_2}{\sin \alpha_1} + 1 \right) \sin \alpha_1} \quad (13)$$

$$S_2 = \frac{P}{\left(\frac{\sin \alpha_1}{\sin \alpha_2} + 1 \right) \sin \alpha_2} \quad (14)$$

The 180 Hp. Mercedes Aero-Engine*

The following report on the design of the new 180 hp Mercedes engine is based on an examination of the engine (No. 20244) taken down the captured German Albatross airplane D.5 A. (1077), which was shot down by anti-aircraft fire in the 5th Stripped zone on November 14, 1917, and the information given by the designer of the engine and the various parts of its general performance, have been compared from records of tests carried out at the Royal Aircraft Factory.

The 180 hp Mercedes engine is the first engine of a new type to be used in service since the advent of the 200 hp Mercedes engine in the early part of last year. These 200 hp engines were apparently so successful that the 180 hp type

Full compression ratio 16 to 1
Oil consumption per h.p. 1 lb.
Air consumption per h.p. 1 lb.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.

Full compression ratio 16 to 1
Oil consumption per h.p. 1 lb.
Air consumption per h.p. 1 lb.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.
Air velocity 100 ft. per sec.

Cylinders—The six separate cylinders are exactly the same construction as those used in the standard 160 hp Mercedes engine, being built up entirely of steel, with the valve pockets turned and milled into the cylinder heads, and the water jackets of pressed sheet steel welded in position. The pistons

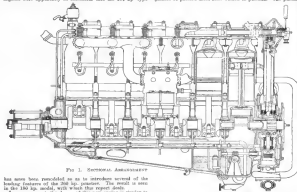


Fig. 1. Sectional, Comparative

has been here, modified as to introduce several of the leading features of the 200 hp pistons. The result is seen in the 180 hp model, with which this report deals.

In many respects the design of this new engine is similar to the 160 hp Mercedes, which is now obsolete. The cylinders are of the same construction and of the same bore and stroke as the 200 hp, that is, 140 mm. x 160 mm., as also are most of the remaining parts; in fact, the engine might well be termed "The New 160 hp Mercedes."

In comparison with the standard type 200 hp Mercedes, the new engine shows a marked improvement, both in the design as a whole and in its general performance during power and economy tests, and as a comparison between the two engines the following comparative table of the leading particulars of the engines is herewith given:

	180 hp	200 hp
Bore	140 mm.	140 mm.
Stroke	160 mm.	160 mm.
Compression ratio	16 to 1	16 to 1
Oil consumption per h.p.	1 lb.	1 lb.
Air consumption per h.p.	1 lb.	1 lb.
Air velocity	100 ft. per sec.	100 ft. per sec.
Air velocity	100 ft. per sec.	100 ft. per sec.
Air velocity	100 ft. per sec.	100 ft. per sec.
Air velocity	100 ft. per sec.	100 ft. per sec.

* Report issued by the Technical Department (Albatross Division) Ministry of Munitions.

† In the Dec. 1, 1917, issue of AVIATION.

‡ This weight is weight of engine (dry), including propeller shaft and exhaust manifold.

structure of the 180-hp engine, also resembles in many ways the 200-hp type. The usual Mercedes practice of casting the lower half of the main bearing housing integral with the lower half of the base flange, and also the method of holding down the cylinder by long bolts which pass through the base flange, top half and secure the two halves of the crank-shaft, is adhered to.

Valve Gear—The angle inlet and exhaust valves of each cylinder, which work at an angle of 15 deg. to the vertical axis of the cylinder, are superimposable as in the 160-hp engine and are of similar design, the valve opening gear is, however, of new design, and follows more the construction of the valve gear in the 160-hp Mercedes engine.



Fig. 2. CRANKSHAFT AND NEW VALVE GEAR, SHOWING PISTON-ROCK SHAFT VALVE REGULATOR AND NEW DESIGN OF VALVE GEAR

General details of this construction and working of the valve gear appear in the sketch, Fig. 2.

It will be noted that the rocker arms and their spindles are now integral, being machined from solid forgings. The camshaft using is constructed entirely from machine-iron castings, and the valve rocker spindles work in direct contact with the machine-iron, no bronze bushes being provided as bearings for the rocker arm spindles, and the crown of the camshaft being formed by the top portion of the rocker spindles housing.

The rocker spindles are hollow, and are lubricated through two holes drilled radially in the spindles by oil thrown off the revolving main shaft the two holes drilled in the rocker arm supplying the oil.

This design of valve gear is undoubtedly a great improvement on the arrangement adopted in the 160-hp Mercedes, the rocker arms of which are working through slots in the camshaft casing, which are provided with felt-packing strips and buffer pins for retaining the oil in the camshaft casing.

Camshaft—The camshaft is of similar design to the 160-hp



Fig. 3. DRIVE SIDE OF ENGINE

Mercedes, and the timing is supported on long studs which are secured into the head of each cylinder.

Wank angle of the valve stems, this, it will be noticed, is different from the standard 200-hp. Mercedes, as shown in the comparative list of leading particulars. The valve lift has been increased from 5.480 in. to 5.180-hp engine to 5.602 in. Only very minor differences occur in the actual dimensions of the half compression gear in the new engine.

Carburetors—No alteration has been made in the design of the vertical dual carburetors. Such carburetors are contained in a cast aluminum water-jacket, which is coupled at the bottom by a water-pipe to the delivery pipe of the water-cooling and at the top to the water-jacket of the lower cylinder in the top portion of the water-jacket of the carburetors, as shown in Fig. 2.

The carburetor feeds three cylinders through a branched induction pipe of steel coil, which is lapped with asbestos and bound with asbestos tape. The throttle are, of course, interconnected and are operated by a cable and also by a

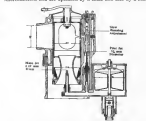


Fig. 4. BURNER OR CARBURETOR

rod lever and rod. The first chamber is of auxiliary design, but is fitted with separate filter attached to the bottom of each air chamber, which are easily detachable. These filters are provided with needle-valve drain cocks. No compensating carburetor. A semi-circular sectional view of one of these carburetors is reproduced in Fig. 5, for reference.

The bore of the main jets is 1.675 mm. and the bore of the pilot jets .685 mm., which is the same as in the 160 hp Mercedes carburetor. A semi-circular sectional view of one of these carburetors is reproduced in Fig. 5, for reference.

The water-jacket in the carburetors is taken through the passage cut in the center portion of the top and bottom halves of the base-flange, which forms an air chamber between the front and rear oil pumps in the lower portion of the bottom half of the base. Air enters the central air chamber through two holes cut in the sides of the chamber and also works air through a large diameter pipe leading from the central portion of the top half of the crank-shaft.

Crankshaft Drive—The method of driving the camshaft through the crankshaft is shown in the sketch, Fig. 3, in a vertical section. This method is similar to that adopted in the 200-hp engine, and is so designed as to allow of a certain amount of vertical adjustment of the crankshaft. The driving end of the vertical shaft is machined and ground in a key-way in the driving flange a ground taper on the bottom extension of the lower, which is 5.020 by 5.000 mm. and, in which, when worn a ring nut, locks the lower securely in position.

In the old 160-hp Mercedes engine the crankshaft driving end on the vertical shaft is fixed by two bolts in the right half of the head, which is fixed in a ground taper on the vertical shaft.

Referring to the sketch of the vertical shaft, it will be seen that the bottom end of the shaft is entered on a "Shifter" or "Reverse" bearing, which is mounted on a steel sleeve, carried at the rear end of the crankcase, and is driven off the sliding bevel gear on the rear end of the crankshaft in the usual way.



deep. With the original hydrogen chamber of 29 mm depth the larvae did not reach its maximum permeability until several hours after starting the test, even though the rate of passage of the hydrogen was several times the normal rate. With the hydrogen space restricted to a constant permeability was reached 30 minutes after starting the test.

The cells are normally used in a vertical position. To determine whether this had any effect on the result, a cell was operated at various other positions. As was to be expected, no effect due to position was observed.

In a similar manner the effect of direction of gas flow was tested. With a cell suspended in a vertical position it made no difference in the apparent permeability whether the air was introduced at the top or at the bottom of the cell. In opinion it is our practice to introduce the hydrogen into the cell at the top and the air at the bottom, on the theory that the cell will be swept out more rapidly. It probably makes little, if any, difference, however, if the resistances are small and the gas current sufficiently rapid.

To determine what effect the area of the test piece had on the apparent permeability, coils of different areas were constructed. These coils had fabric areas of 50, 100, 250 and 1000 cm².

If the hydrogen which passes through the fastest is not rapidly removed from the surface, the partial pressure of the hydrogen might become sufficiently great to lower the apparent permeability appreciably. The lower the average partial pressure of hydrogen in the sample, the faster the gas will grow out of the sample at a given rate of diffusion. The accumulation of hydrogen in the cell is prevented by the current of air which reduces the average concentration of hydrogen in the cell to 1 part per ten thousand. For these reasons it is to be expected that the apparent permeability would be independent of the size of the sample and of the area of exposed fabric within reason.

The results of a series of tests with these cells showed that these membranes were substantially correct. The small (50 cm²) cell gave numerically the same apparent permeability as the 250 cm² cell, but the results were less consistent. The agreement between the 100 and 250 cm² cells was very good. With the 1000 cm² cell apparently a longer time was required to reach equilibrium.

If the hydrophobe is able to penetrate intensely between the plates, the effective area of exposed surface may not be defined by the area of the plates, but by the area of the hydrophobe. This effect would tend to cause the proportionately smaller small oil cells to be a large one. No such consistent difference was noted. It has been found, however, that where a heavy coating of hydrophobe is present, the hydrophobicity and the structure of the fabric, or where a light cloth is used, but the rubber does not penetrate between the threads (and such extra damage occurs), the resulting apparent permeability is fairly constant, regardless of the area of the hydrophobe. The longer the area, the greater the area of the gas chamber. If the fabric has no rubber on one of the sides, and the proper application of waxing to the margin will seal the fabric and prevent leakage on both sides, the waxing material penetrates the rubber and seals the fabric. The best way of reducing the error from the waxing in the second case is to reduce the marginal area to a negligible amount.

(13) *Construction of Hydrogen*—The facts discussed so far were made with hydrogen of approximately 99.9 per cent purity. If the reconstruction of the hydrogen is omitted with the failure be reduced by adjustment of air, it seems probable that the apparent permeability of the failure to hydrogen will be reduced about proportionally. Bury and Thomas¹¹ conclude from two experiments with mixtures of air and hydrogen that the permeability is directly proportional to the partial pressure of the hydrogen.

No tests of this point have been made as yet at the Bureau, since for test 12, purposes the use of pure hydrogen is desirable from every standpoint, and it is known that if the hydrogen contains air, the apparent permeability to hydrogen will be reduced about proportionally. This is an important point to remember in connection with vacuum-loss methods, since the diffusion of air into the hydrogen chamber will lower the partial pressure of the hydrogen and consequently the apparent permeability.

¹¹ August Gonsi, *Belgian Committee on Antisemitism*, 4 p. 776, 1916.

c. Volume-Force Method

[illegible]

The pressure is considerably larger [Goussard et al., 1986]. The considerable reduction of the partial pressure of the hydrogen through the small volume of sea which diffuses away. Attached to the hydrogen chamber is a graduated measuring device which allows one to determine the change in a constant-pressure flask. As hydrogen escapes through the tube, water from the constant-pressure flask drops into the burette through the orifice as held in place. The decrease in the level of the liquid in the burette indicates the increase in volume of water in the bath. The total volume of the apparatus is known and corrections are applied for the changes in temperature and pressure which may occur during the test. The accuracy of the measurements depends on the density of the liquid; further correction is necessary for the change of permeability of the tubes with change of temperature.

In experiments there was oxygenation (20%) between the well-aerated layers of the soil. The gas chambers were then sealed for tests by placing it under a vacuum pressure of air. If it was air-tight, the air in the chamber and the bacteria hydrolyzed soil with hydrocarbon in the chamber. The hydrocarbon was then removed of 50 mm of water. The decrease in soil volume was then noted at 30-minute intervals. A current of air was passed through the upper chamber during the period of test.

This apparatus is similar in principle to the apparatus of Lebedev, Goltzman, and others and examines their best features. It was developed primarily for experimental purposes.

(b) Results of Tests.—The results obtained with this apparatus, using a series of fibers from three different man-made sources, are shown in Table 2. Five purposes of comparison the permeabilities of the same fibers as determined with the apparatus shown in Fig. 3 are included.

The following data from these results and the experiments on the use of this and similar apparatus may be summarized as follows:

TABLE 2.—APPARENT PERMEABILITY AS DETERMINED BY THE
100- μ M LINE METHOD

Fabric %	Supporting responsibility on all 6 factors volunteers used method lowest per program member per hour living	Preparation on all 6 factors programmers used method lowest per program member per hour living	Ratio of preparations
95/5	1.7	1.6	0.94
90/10	1.7	1.9	0.90
85/15	1.6	1.7	0.97
80/20	1.6	1.7	0.97
75/25	1.6	1.7	0.97
70/30	1.6	1.7	0.97
65/35	1.6	1.6	1.00
60/40	1.6	1.6	1.00
55/45	1.6	1.6	1.00

With this use of cell the actual volume of hydrogen escaping per hour is usually small and the errors of measurement relatively large. The results are not very concordant and the precision secured is low compared with the standard apparatus.

Great care must be taken to secure a gas-tight joint between the fabric and the cell in order to prevent any leakage other than through the fabric. A small leak in the hydrogen chamber of apparatus of the protonium type is disastrous, but

with the volume-free method it is fatal to accuracy. Even if this point is negligible, hydroxy acid, which leaves the cell by being in equilibrium with the plasma of the tubule, is a true permeant molecule subjected to trans- and counter-transport.

Since¹¹ to guard against this, if possible, the edges of the tubule were sealed with cellophane. Where a high permeability is indicated, the results are usually open to the suspicion of being influenced by a leak.

1.3. Relation of Volume-Fraction Permeation Method.—To determine the theoretical relation between the volume-fraction and permeation methods it is necessary to know the relative permeability of balloon rubbers to air and to hydrogen. The relative rates of penetration of rubber to nitrogen, oxygen, and hydrogen are 1:2.56 and 5.6. Since the permeability is proportional to the partial pressure, the relative permeability to hydrogen can be written as 5.6

$$w = 5.6 \times \frac{p_{H_2}}{p_{N_2} + p_{O_2} + p_{H_2}} \quad \text{where } w \text{ is at } 1 \text{ atm}$$

to 1. These figures of relative $\frac{p_{H_2}}{p_{N_2} + p_{O_2} + p_{H_2}}$ for different rubber compounds and different experimental conditions

The rates at which hydrogen and air pressure in a balloon fiber have been determined simultaneously by using two mass flowmeters. With the standard cell (Fig. 1) the hydrogen passing into the air stream was determined at the usual meter. The air passing into the hydrogen in the other chamber of the cell was determined at the same time in another meter foremeter by comparison with hydrogen from the same source which had not passed through the cell. The calibration of the aftermeter was necessary in order in this latter case to

参考文献: [1] 王德胜, 王德胜, 王德胜. 2013. 中国人口老龄化与人口红利. 人口研究, 37(1): 1-10.

Faktor No	Persentase ke-40-an dalam 100 dari 100	Persentase ke-40-an dalam 100 dari 100	Persentase ke-40-an dalam 100 dari 100
1000	100	100	100
1000	100	100	100
1000	100	100	100

to the fact that the "air" in the hydrogen may have a different ratio of oxygen to nitrogen than in atmospheric air (see Section 2), but if the oxygen content was as high as 10 per cent, the readings would be approximately



Pre 5 Volume Lo
Announcer

¹⁰ Anthony M. Iritani, *Continuously Permitted and Its Siblings: The Effect of the Family and Not at the Edge of the Law*, 1997.

permeability determined by the penetration method. This value is very close to the one found in the author. The volume-loss results given in the table were on the average 60 per cent of the values obtained with the same factors by the permeation method. The difference of 40 per cent and the value (14 per cent) just calculated must be attributed to the fact that the hydrogels were saturated with water vapor in the aluminum components and was used dry in the other method. It was shown in Section 3b that the apparent permeability was 4 to 5 per cent lower when moist test was used.

(4) Absorption of hydrogen by Fisher—Fig. 1 shows curves with the tests of the apparatus shown in Fig. 7 a series of determinations were made of the rate at which hydrogen was absorbed by the rubber in a ballroom flask. The tests were made by placing a rubber in the apparatus as for a determination of permeability, and then cutting aside in the flask as soon as the test was completed. The flask was then filled with hydrogen, closed, and the rate at which the hydrogen was absorbed was measured in the usual way. The hydrogen absorbed was calculated in terms of volume at normal pressure per 24 hours. The results are shown in Table 4. A comparison of the results with those obtained by the absorption of hydrogen, thus proving what was evident—that the hydrogen was absorbed by the rubber and not by the resin.

TABLE 4.—ABSORPTION OF NITROGEN BY RUBBER-COATED BALLOON FABRIC

[illegible]

These experiments indicate that saturation of the rubber with hydrogen is practically complete in 30 to 60 seconds. Kauer¹ also found, using another method, that absorption was complete in about 1 hour. No further absorption was noted by later experiments lasting 3 to 5 days. It is therefore assumed by the volume-loss method shown in Table 2 that the reaction at the end of the first 30-second period was usually about 10% higher than the average of the 3 to 5 16 readings taken. Although the data are not very complete, it seems evident that a higher initial reading would be partly due to the absorption of hydrogen by the rubber, and partly due to the significant difference in the results because of the rather large accidental errors involved in the volume-loss measurements.

(c) *Conclusions*—It may be concluded from the results of these tests and many others unreported and from the experience of other experimenters as noted in the literature that a volumetric method as far developed are inferior to the potentiometric method both in precision and reliability. The permeability according to the volumetric method is approximately 70 per cent of the permeability by the potentiometric method.

*Standard To

The permeability test, adopted as standard by the Bureau of Standards, may be broadly described as follows. The fabric to be tested is held in a permeability cell of the size and shape shown in Fig. 4, or its equivalent, and a maintained differential of test air at a temperature of 25°C. & a current of pure dry hydrogen is passed in or out of the fabric at a rate of a quart or of one liter of water above the pressure on the other side of the fabric. Dry air, at approximately atmospheric pressure, is passed at a constant rate through the other chamber of the cell and the hydrogen passing through the fabric into the air is determined by means of a gas difference manometer. The hydrogen flow is measured by a gas flowmeter and weighed as water. If the anemometer is used

¹² Field, *Analystes des Fats*, 1860 Chem. 44 p. 347; 1869

a reading is taken every half hour until a constant permeability is achieved. If the condensation method is used, the fabric should remain in the apparatus with the closed valves of pure hydrogen for a period of one hour before beginning a test.

The permeability is calculated as liters of dry hydrogen absorbed per square meter of fabric per 24 hours, the volume of hydrogen being converted to the standard conditions of 0° C and 760 mm. pressure of mercury.

B—Operating Directions and Calculations

In the preceding sections many of the details of operation such as pressure, temperature, etc., have been described at length. It may be well to call attention also to certain other points which must be taken account of in order to secure accurate results.

To begin with the fabric should be firmly fastened to the cell and the area of fabric exposed to the hydrogen should be definitely known. If the faces of the cell are exactly square or if they are satisfactorily laminated so that the cell is not deformed under high pressure, hydrogen may leak past the edge of the gas chamber and increase the area of the exposed fabric. The leaks will necessarily be too high and will probably be erratic.

The air current should be maintained as uniform as possible. The air can be forced through under low pressure at the desired flow rate through slightly reducing the pressure in the outlet with a water pump or other vacuum pump. To aid in maintaining a uniform pressure, a water-gas, gas-pressure regulator may be used. A long piece of ordinary glass tubing inserted in the air line also tends to reduce fluctuations in the rate of air flow.

The hydrogen should be passed through the cell rapidly at the start of a test in order to sweep out the air as quickly as possible. When the air is removed the hydrogen used be passed rapidly through the cell in order to sweep out the air which diffuses through the fabric.

The permeability is calculated from the following equation:

$$P = \frac{V \times F \times N}{A}$$

P = permeability in liters per square meter per 24 hours
 V = rate of air passage, liters per 24 hours.
 A = area of fabric in square meters
 F = factor by which indicated air rate must be multiplied to reduce the gas volume from the condition of saturation to the temperature and pressure in the meter to the volume when dry and at 0° C and 760 mm. pressure

N = percentage of hydrogen in the air stream.

The factor F is calculated as follows:

$$F = \frac{P - P_v}{760} \times \frac{273}{1 + 273}$$

P = barometric pressure plus pressure above atmosphere in the meter
 P_v = temperature of meter
 P_v = vapor pressure of water at temperature t

A table giving the value of F for different temperatures and pressures should be prepared if many tests are to be made.

Most of the methods now discussed in sections 5 and 6 apply equally well to a method in which the hydrogen is determined by condensation with subsequent weighing as water retained by means of an anhydrometer. The advantages the anhydrometer offers in point of speed and convenience have already been pointed out. If a suitable anhydrometer is not available, however, the results may be determined by condensation. In fact, this is the customary method in commercial use.

To secure correct results by condensation, it is necessary that the air and hydrogen from the cell be perfectly dried; that the hydrogen be completely removed; that the water from the cell be completely absorbed and its weight correctly determined. This can be done with sufficient accuracy if the proper precautions are taken.

The efficiency of drying with any apparatus and drying agent should be tested by blank runs. It is desirable to use two absorption tubes in series in order to determine the efficiency of drying as well as the amount of water. The first begins to absorb increasing amounts of moisture, the first tube should be rebilled.

The condensation of the hydrogen may be accomplished in a number of ways, such as by passing over heated copper oxide, phosphorus pentoxide, phosphorus pentoxide, phosphorus pentoxide, phosphorus pentoxide, etc. Complete condensation can be secured by any of these methods if the proper temperature is employed and the gas is not passed through the condensing tube too rapidly. In any case it is well to determine by experiment that the condensation is complete under the conditions of use. It is important that the rate of passage of the air be not too rapid to prevent efficient drying and complete condensation.

Since the amount of water collected is usually small, every precaution should be taken in weighing the gas absorption tubes. It is desirable that a measurement of as nearly the same size and shape as the absorption tube be used in weighing. The absorption tubes should be weighed with a clean cloth and hung in the balance case or a suitable container to prevent drafts to increase the temperature of the balance before weighing. Rubbing the glass while dry is likely to produce the electrostatic charges on the glass which prevent accurate weighing.

C—Accuracy of Methods

The accuracy with which the permeability of a fabric can be determined depends upon the accuracy with which the various factors of the test, such as concentration of hydrogen, etc., can be determined, and is closely associated with the question of uniformity of fabric. Some fabrics show very close agreement on duplicate tests and give evidence of being very uniform. Others show 24% to 30% variation in the results of single different test pieces of the fabric showed permeabilities between 8.5 and 9.5 liters. Experiments with the fabric which show that a great number of the fabrics are caused by real variations in the permeability of the fabric and not by unbalanced strains of testing. If the fabric contains "pockets," the variations in permeability of test pieces may be very large because of its probability that the small holes will be uniformly distributed and be uniform in size. One sample which showed "pockets" gave tests ranging from 14.5 to 43.5 liters. In general, duplicate tests on good fabrics show an agreement within approximately 1 liter in the range of 18 to 20 liters.

With the standard cell having an area of 300 sq. m., an anhydrometer giving the concentration of hydrogen to 0.01 per cent and a U-tube water meter graduated in thousandths of a cubic foot and read to ten thousandths of a cubic foot, it is possible to have the probable experimental error below 5 per cent. Under routine testing conditions the accuracy may vary from 5 to 14 per cent. Greater variations than this can be the result of the fabric or the conditions, and will. The practice attained—that is, the agreement between successive tests on the same test piece—was somewhat greater than the absolute accuracy attained.

If the condensation method is used, about the same accuracy can be secured if the proper precautions to secure complete drying, condensation, and absorption are taken. However, the length of time necessary for a test, check tests are not usually run, and furthermore it is necessary to obtain the average permeability over a period of several hours unless a very large cell is used.

D—Summary

The various methods for determining the permeability of fabrics exposed to hydrogen are described in this paper, the procedure and rapid values and the factors of similarity are described in detail. The phenomenon of the passage of gas through rubbers by solution in the rubber is discussed in relation to testing methods. Data are given to show the effect upon the apparent permeability of different experimental conditions such as temperature, pressure, humidity of the gas, duration of test, etc. A knowledge of the effect of these factors enables one to compare results of tests made under different conditions. It has been shown that methods which depend upon the loss of volume of hydrogen confined by a sample of the fabric give results which are about 70 per cent of those obtained by the standard method which measures the volume of hydrogen penetrating the fabric. This is due to the actual permeability of the fabric to air and hydrogen.

The author is indebted to Messrs. Francis A. Smith, P. & LeRoy, S. F. Pukanzig, and L. L. Moore for their assistance in making the measurements reported.



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Facts About Airplane Construction

Another story is going the rounds of the daily press that the aircraft program has struck a new snag. The gist of this story is to the effect that the DH-4 is not a success, and that General Peckham recently notified regarding that his own share of this type of machine made in France.

Another version is that General Peckham has recommended that further shipments be held up until certain alleged defects shall have been corrected. The War Department is quoted as the source of this information, and the Signal Intelligence Committee on Military Affairs is also said to have made a thorough investigation of DH-4 construction and will discuss not a little again in the subject in its forthcoming report.

The first version of the story, as above outlined, is absolutely denied by Mr. John William S. Potter, Assistant Director of Aircraft Production, and in contrast to it Mr. Potter read to the Washington representative of AVIATION and AVIATIONER, concerning a telegram from General Peckham, dated July 24, that he received on July 25. In this cable orders were given for the prompt shipment of planes, engines, etc., on August 1, and DH-4s named at the head of the list. Mr. Potter is emphatic in his refusal to be influenced by any such story.

On the other hand, to quote an old adage, "Where there is smoke there is fire." It is admitted that certain additions, or so-called improvements, have been made in the DH-4 is more foreign ideas, and what is more important, that DH-4 production will eventually be abandoned and the DH-5 and DH-6 will be put into production in its place in this country. These are types which have found great favor in England. Just when the change will be made is not known. It is possible that the entire number of DH-4s called for by the air program may be manufactured, or it is as feasible to discontinue their production meanwhile, that that will be done. Who cannot well depend upon circumstances, but it is gathered that the latter is the more probable.

Incidentally, it may be mentioned that many reports have also been made to the effect that the RE-6s are being built in place of the Bristol fighters, whose production was recently ordered discontinued. This may be satisfactorily stated as erroneous, for the very simple reason that this type was and is a part of the original air program.

Loening's Monoplane Has Test Flight

The first flight of a two-seater biplane monoplane, designed and built by Grover C. Loening, the aeronautical engineer and constructor, at his factory in Long Island City, was made at the Army Aviation Field at Mineola, August 5. While details of the construction of the machine and the flight are not available for publication, it is claimed that the construction is such that the plane has an unusually large radius of vision and the engine has practically no "blind spots." It is stated that the unique design of the machine has overcome the well-known drawbacks of the monoplane in strength and range and that its engine machine showed low working speed and exceptionally good stability.

It is claimed that the simple design of this machine permits an construction with about one-tenth of the number of parts of the ordinary machine. It is powered with a 200 horse-power Hispano-Suiza engine.

Aircraft Production Assistant

C. W. Nash, president of the Nash Motors Co., Kenosha, Wis., and formerly president of the General Motors Co., has been appointed by John D. Ryan, Director of Aircraft Production, as assistant to the Director in charge of manufacturing and production. He will assume his duties immediately.

This appointment does not change in any way the organization already effected in the Bureau of Aircraft Production except to give Mr. Ryan an additional assistant.

But Morris Airplane School

The But Morris Airplane School, Lincoln Avenue, Chicago, Ill., has received much of the country's flying talent. The school represents a graduate of the school. Four of the graduates are now instructors at the Great Lakes Training Station where they are in the Air Service. The course at this school is eight weeks.

Naval Aircraft Factory

July 27 was the anniversary of the date on which the building of the Naval Aircraft Factory at Philadelphia was commenced, and on that day Rear Admiral David W. Taylor, chief of the Bureau of Construction and Repair, which has and operates it, reported to the Secretary of the Navy his satisfactory record made in its erection and operation.

In recognition of this record Secretary Daniels addressed the following letter to Naval Constructor F. G. Collins, manager of the plant:

"The Department desires to express its appreciation of your ability shown in carrying out the duties of the Naval Aircraft Factory, and in bringing it, as its manager, to its present state of efficiency. One year ago the construction of the Naval Aircraft Factory was authorized by the Department, and you were charged with its management. The factory has been built in accordance with plans prepared by you, and the records show that from the first flying boat were built October 15, while the building was not completed until November 28. The first flying boat was given its successful first flight on March 21, 1916, and since that date a steadily increasing rate of production has been maintained. It is noted that the first order for fifty large flying boats has been completed and the greater part are now flying over British waters."

The contract for the aircraft factory was awarded August 6, 1915, and work was begun on the same day. The original factory had a floor space of 169,000 sq. ft. All extensions, which will give an added space of 154,000 sq. ft., was begun on February 23, and is now positively completed.

Aeronautical Commission to Go Ahead

Charles H. Wilford, chief engineer of the Aeronautics Plant and Motor Company, has been given leave of absence by his company, so that he can join a Government aeronautical commission to be formed in France and England. It is understood that the purpose of the trip is to secure the latest military information upon aircraft and to translate it into production for next year's American airplane program. Colonel Hall and Major Hoffman, of the Aircraft Production Bureau, are also members of the delegation.

Headlines Malicious

One of the principal reports of Headlines is malicious, and early on it is given in the United States, says R. E. Johnston, a Massachusetts, who has lived in that country for his past twelve years. He comments:

"There has been an extraordinary demand from the British Government for the malice, and I presume William D. H. will be sent to this country within the next year, all of which will be controlled by the British Government."

Goodyear Gets Airship Contract

Among the July contracts placed by the Bureau of Supplies and Accounts of the United States Navy was a contract for airships with the Goodyear Tire & Rubber Co., Akron, Ohio, and a contract for construction made with the Chapin-Kearney Co., Cambridge, Mass.

Following were among the July contracts placed by the General Engineer Office of the Army: The Johnson & Mann Dairyming Co., aircraft tire cases; Fox Production Book, Washington, D. C., air, Chapman Spark Plug Co., Toledo, Ohio, with plans; Sperry Gyroscope Co., Brooklyn, N. Y., compasses; and Riedelhof Electrical Co., Newark, N. J., navigation and instruments.

Changes in Rich Tool Co.

Bernard G. Johnson has joined the sales department of the Rich Tool Co., Chicago, Ill. He will make his headquarters at the Detroit office, 709 Kuper Park.

On July 1, J. Henry Shaw, manager of sales, Motor Tool Department will make his headquarters in Chicago, where all sales will be handled. Mr. Shaw has been in Detroit for his past three years.

Auto Course at Michigan Auto School

The Michigan State Auto School, Detroit, Mich., is not giving a course in aeronautical engineering. The students are taught the assembly of engines and planes.



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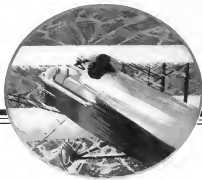
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Aluminum Draining Roller Co.	123	Leygraf & Co.	15	
Aluminum Castings Co.	81	Leuzing, Greiner Co.	15	
Aluminum Co. of America	122			
American Propeller & Manufacturing Co.	133	M		
Automotive Engineering Co.	133	Martin, Glenn L. Co.	7	
B		Mattison, C., Machine Works.	15	
Baker Caster Oil Co.	120	Mays Radiator Division	8	
Baroids Mfg. Co.	121	Motor Engineering Co.	8	
Bell, Julius, & Co.	122			
Boston Auto Gauge Co.	122	N		
Brook, Arthur, Jr.	122	National Wire Wheel Works	15	
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Burgess Co., The	36	New York Wire & Spring Co.	15	
		Norris Co. of America	7	
C				
Cable Light Manufacturing Co.	125	O		
Continental Aircraft Corp.	112	O'Donnell Engineering Co.	15	
Curtiss Aeroplane & Motor Corp.	111			
D		P		
D. W. F. Engineering Co.	131	Palmer Stamping Co.	15	
Dallas Manufacturing Corp.	124	Park Deep Forge Co.	11	
Dallas Metal Products Co.	127	Poss, Flexible Metals Tinting Co.	11	
DeLuxe Machine Works	124	Pratt & Lambert	15	
Dixon, Joseph, Crucible Co.	127	Pyral, F., Machine Co.	15	
Dixie Die Casting Co.	135			
Dodge, W. A.	134	R		
Doring Systems, Inc.	128	Radium Dial Co.	11	
Dunsmuir Machine Corp.	128	Radium Locomotive Material Corp.	15	
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E		S		
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Fedders Mfg. Co.	83	Stinson, Edgar B. Co.	15	
Ferdinand, L. W. & Co.	135	Strains & Bangslev	15	
Fine Manufacturing Co.	122	T		
Fosberg, The Co.	131	Taylor Instrument Companies	15	
Fox Machine Co.	124	Thomson-Morse Aircraft Corp.	15	
		Tops Steel & Iron Co.	15	
G		U		
Gillette Safety Razor Co.	96	Union Gas Engine Co.	15	
Goodyear Tire & Rubber Co.	87			
Grand Rapids Veneer Works	117	W		
G and O. Mfg. Co.	120	Wabco Mfg. Co. & Co.	15	
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